US ERA ARCHIVE DOCUMENT

## 104 EPA STAR Graduate Fellowship Conference

**Next Generation Scientists—Next Opportunities** 

# onitoring Mercury Transport in Estuarine Sediments Using Novel Reactive Membranes

## **Background**

tudies of contaminant transport in the coastal zone emonstrate that mercury (Hg) is effectively trapped nd recycled within estuary sediments.

Inderstanding the dynamics of Hg ransport and bio-availability

vithin estuarine sediments is mportant as porewater (i.e., vater held within the sediment natrix) exposure may play a ignificant role in the Hg ptake potential of sediment-

welling organisms. As these rganisms often exist near the ase of coastal marine food chains.

lg exposure within sediment orewater may facilitate Hg ransfer throughout the food web.

ID SELD FOR DOOR - ZCX ANSO

chitosan membrane ~ 5 mm.

Water Hg-specific membrane

Proposed field deployment strategy.

Total thickness of Teflon holder +

Sediment verter A Hg A Hg

COH Hg A CHH

Cross-sectional view of scaffold

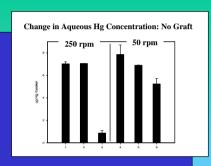
reinforced chitosan membrane.

Scale bar = 1 mm.

#### **Deployment**

Chitosan membranes are deployed within estuary sediments to assess concentration and transport potential of Hg within sediment porewater. By matching tidal stage (rising or falling) to deployment time, the extent to which the resulting geochemical changes affect mobilization of porewater Hg species may be studied.

Reactive membranes may ultimately function as assessment tools for the remediation of contaminated sediments.



Treatments 1 & 4 = time 0; Treatments 2 & 5 = time 6 hrs for controls containing Teflon holders w/o membranes; Treatments 3 & 6 = time 6 hrs for Teflon holders w/ chitosan membranes.

chitosan membranes.

## Objective

he creation of a macroporous, tabilized porewater sampling tembrane based initially on hitosan, a derivative of crab hells that has been chemically nd physically modified to

nhance affinity and uptake capacity for Hg.

95KV X266 1680 881 10213 UM

Plan view of macroporous chitosan.
Scale bar = 100 µm

Effect of Grafting on Membrane Hg Uptake

1.0

1.0

0.4

0.4

0.4

0.7

NG-No-OM NG-OM CG: No-OM TG: OM TG: No-OM TG: OM

NG = no graft; CG = cysteine graft; TG = thiourea graft; OM = organic matter (5 ppm humic acid); 6 hr exposure @ 50 rpm;

Results with matching stars (\* or \*\*) statistically equivalent at p <0.05; data as mean ± 1 s.d. (n = 6).

## Research

Uptake at pH = 7.0; 0.01 M KNO<sub>3</sub>; 0.1  $\mu$ M Hg (as Hg(NO<sub>3</sub>)<sub>2</sub>); 6 hr exposure; data as mean  $\pm$  1 s.d. (n = 6).

Initial research has focused on Hg uptake:

- •By sulfur (-SH) and amino (-NH<sub>2</sub>) grafted versus ungrafted chitosan membranes
- •In the presence versus absence of dissolved organic matter
- •Under varying agitation rate

We are currently testing membrane performance in sediments of known Hg contaminant level.

